

13

ture with the metal oxide substantially uniformly distributed throughout the nanoarchitecture.

7. The nanocomposite material of claim 1 wherein the graphene layer comprises functionalized graphene sheets.

8. The nanocomposite material of claim 1 wherein the graphene layer consists essentially of 6 to 29 graphene sheets.

9. The nanocomposite material of claim 8 wherein the graphene layer comprises functionalized graphene sheets.

10. The nanocomposite material of claim 4 wherein the titania is in a mesoporous form.

11. The nanocomposite material of claim 4 wherein the mesoporous titania is in a rutile crystalline structure.

12. A method comprising:

providing graphene layers in a first mixture, the graphene layers having a first surface and a second surface and thicknesses of 0.5 to 50 nm;

dispersing the graphene layers with a surfactant;

adding a metal oxide precursor to said dispersed graphene layers to form a second mixture;

14

precipitating the metal oxide from the second mixture on surfaces of the dispersed graphene layers to form a nanocomposite material comprising a metal oxide bonded directly to the first and second surfaces of a graphene layer; and

wherein the precipitating the metal oxide further comprises condensing the metal oxide at a temperature of less than 100° C. to form rutile crystalline metal oxide bonded directly to the first and second surfaces of the graphene layer.

13. The method of claim 12 wherein the metal oxide comprises tin oxide.

14. The method of claim 12 wherein the metal oxide comprises titania.

15. The method of claim 14 wherein the titania is mesoporous.

* * * * *